https://doi.org/10.29070/r6tgjj74

Causes of Water Pollution and it's effects and control

Gyanendra Mani Triphathi*

Research Scholar (Chemistry), Magadh University, Bodhgaya, Bihar, India

E-mail: gyanendratripathi927@gmail.com

Abstract - The study focused on "Causes of Water Pollution and its Effects and Control." A sample is a tow sample, which is a sample collected during a specific space movement short time duration period on the effectiveness. Record samples separately at selected locations, depths, and times. The combination of depths is recorded over to inside deep on the water column specific so, either by prior determination on water-column or at a selected place also time duration on the water-column. Biochemical oxygen demand (BOD) is also measured using a pH meter. Chemicals such as sulfates, nitrates, and chlorides have been reported to cause water bodies such as lakes, rivers, and ponds to become acidic.

Keywords: Water pollution, Environment effect, Environment control, and pH meters etc.

INTRODUCTION

A world without wealth creates problems for our future. Ironically, only a small fraction of the world's water actually exists. 97.00% in global that water is much salty for drinking or farming also. And over of some water is stored by glaciers or rest inaccessible places. Only about 1% of the world's water is left as liquid, potable water; Whereas 98% of the water is going underground. In the long term, India requires a renewable water supply of 2000 cubic meters per capita to survive well, while the country represents a water deficit of 1000-2000 cubic meters per year, and 500 cubic meters per year. There is water shortage every year. The pH meters have to represents water scarcity per person per year.

Currently, an additional water supply of approximately 7000 cubic-meters each people on every year is available, providing enough water for three times the current population. However, the study analysis doesn't to intake highly amount distributing also accessibly issues. Current technology desalination; however, this tool on ones suitable for rich coastally areas. The cost of converting contaminated water into good water is very expensive: Transportation of clean water: areas with abundant water.1 With this in mind, the valuable resources of large plastic bags in transportation or wastewater should be properly managed to become a good resource and support people's future.

This management is more important because the country will be able to distribute fresh water to the world. While much of the rest of the world and many countries in the south are experiencing water shortages, other countries continue to use water without any concerns. According to WHO statistics, in the US alone, domestic water use is 291 billion cubic meters, domestic water use is 35.80 billion's cubic-meters, and agricultural water gran is 120.90 billion's cubic-meters. n India, this water does not include water used for other purposes, such as irrigation of crops and for drinking: In our country requires water more than 170 billion's gallons of recycled-water. If seen globally, more than 1/6 of the population depends on clean drinking water. These are partly amount of poor quality of groundwater in many areas, which does not provide safe water for human consumption. Poor sanitation and hygiene can also lead to health problems. Finally, the availability of clean water is a worldwide problem.2

Good drinking water is colorless, odorless, and tasteless. No turbidity or other impurities should be removed. Its pH should be 7 to 8.5 pH levels. It will be not viruses, bacteria, disease-causing organisms. It will be not heaviness metals like lead, pesticides, and other toxic substances. Drinking water should not contain metals such as mercury, chromium, cadmium, cobalt and nickel. It should be slightly soft, preferably between 50-100 ppm hardness. Its hardness should not be higher than 150 ppm. (Love).3 It should be wonderful. It should not stain clothes. It should not corrode pipes or cause calcification.

SAMPLE COLLECTION

There are generally two types of sampling technique. Individual and group samples are samples that are specific in nature and are collected by the method in

a short period of time i.e. few seconds/minutes. Such samples appear to be a permanent and clear representation of the area and its location through a "snapshot" of time. Both the different samples are a selective material in themselves and they are measured in depth over time. Both internal samples are taken over defined portions of a water column, collected from a given reservoir and at a selected location or time.

There are generally two types of models. A downstream sample is a sample collected when a specific energy source is present for a small duration of time (usually 2nd/min). Therefore, they represents a *snapshot* of the remaining area in space and time. Record samples separately at selected locations, depths and times. A combination of depths is recorded over on entered under ther water column, either by prior determination of the water column or at a selected location and time in a water column.

Mixed models would give that supporting sample of heterogenic matrix where the target analyte concentrations may vary over time/space. Compositing a sample it may be put in combination several samples or by using special materials used as samples. Using continuous sampling, collect the mixed sample (temporary) by periodically mixing the water balance of the mixed water.⁴ Flow-rate composites are collected by continuously adding the value proportional become the flows, and mixied of the water balance of the accumulated water over time, or by collecting at certain intervals.

Advantages of composite samples includes reducing couts on analysed big amounts of data, Whenever the quantity of test samples is limited, the more representative the samples of the heterogeneous matrix, the larger the sample size.

The reduction of a group sample to a lower level of analysis may be considered to be a potential enhancement or weakening of the individual samples' ability to show more or less analytical cognition, by creating interest in the intervention with low and stable interest and by analyzing the same, the possibilities of increasing collection in all the pieces can be reached quickly by an experiment, on the other hand, in case of small samples, in order to save time, special arrangements are made for the distribution and insuring of the samples by the commercial carter, all the facts of the sample are set out in the numerical value.5 In this way a sample is recorded by observing a complete chain of observations from a series of different sources from which water samples are taken for the experiment.

GAP IN RESEARCH

Analytical procedured for assay and determination of various regions are very much need in current scenario. A convenient and quick procedure need to be developed. It is for the determination of the

concentration of pesticides in the technical form and in their formulations.

General Procedure: pH-meters are the widely inpland or instrument like electro-metric arrangement of pH levels. The pH meter is used to calibrate the lead and bond to the pH of water in a high-impedance electrometer, The electrometric determination of the pH of water involves the measurement of the electromotive force in a cell consisting of an indicator electrode and an electrode filled with hydrogen molecules that are resolved by the test electrode.

The chemical oxygen demand (COD) test was also performed. For this matter present in the samples in sulphuric acid solution containing excess of 50.00% potassium dichromate gets oxidised to ammonia by refluxing with water and carbon dioxide, while in excess dichromate the ferrous ammonium sulphate gets automatically converted into standard solution. Biochemical oxygen demand (BOD) was also determined. The principle of BOD determination is to measure the dissolved oxygen content of the sample before and after incubation at 27°C for 3 days. Anoxic or low-oxygen samples been the oxygen present in the substance can be saturated and diluted to a specific concentration by repeatedly diluting the salt water to take limited on there oxidation.6

Dissolved oxygen (DO) content is large amount in oxygen in water. Dissolved oxygen was determined by Winkler's modified iodine azide method. Hardness is caused mainly by dissolved Ca and Mg in water. It can be easily determined by EDTA titration as given below. The ions of magnesium and calcium react on EDTA and solubles solution from the success of the titration on indicate with different colour and it will changing on the Eriochrome Black T indicator. During titration, Ca ions do not react with chrome black T indicator. Only Mg + 2 ions react with the indicator and change its color. Ca ions in the titration solution can be obtained.

RESULTS

We know that without water we can not servives in the world, it is true without water there is no life. From the earth there is 71.00% of world is covere on water, only small portion of that is has use to fresh water. Furthe, a 97.00% of the world's water comes from the oceans, which are too salty to drink. Rest 03.00% has place water. While 02.97% are ice age. Therefore, only .003% to this earth is accessible as moisture, soil, water vapor, hence the water as lake, stream, on different rivers, If the aquifer is withdrawn at a rate that exceeds its return value, the water table will drop. Any pollution that reaches the ground will also be drawn into the groundwater, causing groundwater buildup and disease in nearby waters. Water is said to be polluted when its quality or composition is altered, directly or indirectly, by human activities and becomes unhealthy for any purpose. Water pollution is any physicals, biologicals

Journal of Advances and Scholarly Researches in Allied Education Vol. 21, Issue No. 7, October-2024, ISSN 2230-7540

or chemicals changeed in water quality that affects living organisms or makes water unfit for any purpose.

Water is polluted by peoples activities. Pollutants reach water or groundwater in different ways. A source or source of pollution that is easily identifiable is called a point source. For example, municipal and industrial sewers where pollutants get into the water. Non-point pollutants are pollutants that are not easily identifiable, such as agriculture (from fields, livestock and fields), like acid-rain, and storm-blackening etc.

Causes of water pollution - A causes of water pollution. The organisms cause diseases (pathogens) including bact., viru., proto. and parast. are enter the water through household sewage, human waste and animals, and are responsible for many waterborne diseases. Human waste includes bacterias i.e., E-bac. There is a limit to the exes of oxydeation on water may be preventeg in solution. The concentration of dissolved oxygen (DO) in cold water can be as high as 10 ppm (parts per million), compared to about 200,000 ppm in air. It removes dissolved oxygen from the water.

The oxygen concentration on water are essential on aquatic-life. Dissolved O2 in the water body are below 6-ppm, and its grow up different animals will be stunted. Carbon dioxide enters the water through the air or through the photosynthesis process that occurs during the day in many aquatic plants. Organic waste it may be broken down by aerobic-bacteria. Many bacteria consume the oxygen in the water to break down these waste products. During this process, the quality of water decreases. This can lead to the death of fish and other marine life. Therefore, anaerobic bacteria (organisms that do not need oxygen) start digesting the waste. Their anaerobic respiration produces foul odors and odors that are harmful to human health.¹⁰

Agricultural Chemical Wastes: Water and electricity are politically subsidized in agriculture. This makes waste streams less efficient than better practices such as rivers and streams. It should not be just agricultural practices and agricultural methods that promote efficient use of water. Water loss occurs due to damage and causes water to leak and become saline. Agricultural wastes include fertilizers, pesticides, and possibly pesticides and herbicides that are commonly used on agricultural land. Improper disposal of pesticides through farming and farming results in large amounts of contaminants being introduced into water and soil. Some pesticides. Pesticides contaminate water-bodies by leaching into groundwater through surface plowing, spray drift, rain erosion, and direct spraying of dusts and pesticides into low-lying areas. Most are not biodegradable and can persist on our environmental long period. The substances may reached the peoples body might eating habit and cause biomagnification.

Global climate change also affects water-resources has increased evaporat, regional change on heaviness use, timing/frequency (which affect moor flow),

humidity has small changes in climate can have a large face of water-resources, especially, in arid Vs semi-arid region for south resion.¹¹

The study has to impact with agriculture area, A fresh water has to electricity, leading to fewer water resources and land degradation. India relies on perennial rivers in the Hindu Kush and Himalayas, as well as rainwater, and ice water. The monsoon crisis could cause flooding in the Himalayan basin, as rising temperatures will also increase the snow line, reducing natural water capacity on freshwater resources.

Effect of water pollution on plants and water pollution on plants:

- i. Effects of acid deposition:
- ii. Nutrient deficiency in aquatic ecosystem:
- iii. Effects of organic matter deposition:
- iv. Effects of detergent deposition:
- v. Effects of agricultural chemicals:
- vi. Effects of industrial wastes:
- vii. Effects of silt deposition:
- viii. Effects of oil spillage:
- ix. Effects of thermal pollution:
- x. Phytotoxicity effects on plants:

Detergents used in domestic and commercial use can be washed into water bodies and have a significant effect on plants. Detergent contains a lot of phosphate, which causes a lot of phosphate in the water. Phosphate is absorbed into the plant through the roots or the surface, and enzymes inhibit many enzymes, causing plant growth retardation, root elongation, carbon dioxide fixation, photosynthesis, cation absorption, pollen germination and pollen tube growth, chlorophyll and cell membrane damage, and denaturation. metabolic protein processes. procedures. A thermal power plants affect the aquifer by releasing hot water into it. Heat causes oxygen deficiency, which reduces the activity of decomposers. aerobic As organic matter decomposes less, the availability of nutrients in the water is affected. When temperature increases, enzyme activity is inhibited, which reduces the photosynthesis rate of aquatic plants. As water temperatures increase due to thermal pollution, primary productivity and aquatic plant diversity decrease.

When chemicals contaminate water or soil, plants can absorb these chemicals through their roots. Phytotoxicity occurs when chemicals poison plants. Symptoms of phytotoxicity include stunted growth, seedling death, and dead leaf spots. For example, many believe that mercury poisoning from fish can also affect aquatic plants because mercury

www.ignited.in

accumulates in roots and stems, causing bioaccumulation. ¹³ Mercury levels in food increase when animals eat unhealthy foods.

CONCLUSION

Major challenges to improve water management in India include rainfall variability, uneven distribution of surface water resources, persistent drought, groundwater, water and salt pollution due to overexploitation and dumping of waste, urban waste and livestock management in rural areas as well as in urban areas. Areas In addition, there is partial-treated also un-parlal waste-water on residential areas, industrial are established also aquifers. A sewage treatment plant (STP) should be provided in the area to clean the sewage. Unnecessary use of fertilizers, pesticides and herbicides should be avoided in agriculture and organic farming methods should be used. Riverside agriculture should be limited to preserve riverside vegetation. Follows there relies on different areas of Gaya town there wetre more darty place in our socity, So we concede that we clean our socity. The pollute water in this areas has to dumped containing harmful chemicals should be stopped. Rainwater harvesting should be done to prevent water accumulation in the soil. A human make to awaring this problem also to prevent water and air pollution, humans will have to face the depth of this problem because every human being will have to be aware about pollution.is the first step to preventing water pollution. Therefore, apart from the government, all of us will also have to be aware to stop all types of pollution, otherwise the future will be even more scary.

REFERENCE

- 1. A.K. De, The Environmental Chemistry, 3rd Edition. New Age Int. Ltd., New Delhi (1994).
- 2. C. Tomlin, The Pesticide Manual of Chemistry; 11th Ed. British Crop Protection Council. Blackwell Scientific Publications, Cambridge, Massachusetts (1994).
- 3. A.L. Grube and A. H. Aspelin, Pesticide Industry Usage and Sales: 1996 and 1997 The Market Estimates the Office of the Prevention, The Pesticides and Toxic Substances, U.S. Environmental Protection Agency. 733-R-948-00 L. Washington, DC (1999).
- 4. G. W Ware; Pesticide, 6th Ed. The Thomson Publications, Fresno, California (2000).
- Dow Agro Sciences; Midland, MI (1995). U.S. Environmental Protection Agency. The Office of the Pesticides. TOX Onliners- Vernolate (1994).
- 6. S. M. Khopkar; The Environmental Pollution Analysis, The Wiley Eastern Ltd., New Delhi (1995).
- 7. R.S. Doria, Man, The Development and Environment. The Ashish Publishing House, New Delhi (1991).
- 8. Farm Chemicals Handbook. Meister Publishing Co., Willoughby, Ohio. Vol-85 (1999).

- 9. P. MacCarthy, S.W. Cowling, R.W. Klusman, and IA. Rice; Anal. Chem., 65, 244R (1993).
- D. H. Lee, Environmental Factors in the Respiratory Diseases, Academic Press, New Yark (1992).
- U.S. Srce Rarnulu, Chemistry of Insecticides and Fungicides. Second Ed. Oxford & IBH Publishing Co. Pvt. Ltd., New Delhi (1995). Agricultural
- 12. W.T. Thomson; The Agricultural Chemicals, Book-III, The Miscellaneous Chemicals. The Thomson Publications, Fresno, California (1995).
- 13. A.M. Ware and G.W. Hanrahan, Defending Pesticides III Litigation, 10th Ed. West Group, New Yark (2000).

Corresponding Author

Gyanendra Mani Triphathi*

Research Scholar (Chemistry), Magadh University, Bodhgaya, Bihar, India

E-mail: gyanendratripathi927@gmail.com